### WATER RESEARCH CENTER

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# Ecological Significance of Specific Conductance in Streams – Part 2

#### John K. Jackson, Ph.D. Senior Research Scientist Aquatic Entomologist & Stream Ecologist

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### Terms Describing Salt in Fresh Water

### Salinity (ppm)

Total Dissolved Solids (TDS, ppm)

### Electrical Conductivity or Specific Conductance (µS/cm)

Salt (mg/L) (NaCl, KCl, CaCl<sub>2</sub>, MgCl<sub>2</sub>)

Chloride (mg Cl/L)



## Points to Remember

- Electrical Conductivity/Specific Conductance measures dissolved salt concentration
- SC does not tell you about specific salts (NaCl vs MgCl<sub>2</sub>)
- > SC naturally varies  $100 200 \mu$ S/cm in response to precipitation, and  $500 1000 \mu$ S/cm with a region
- Natural <u>spatial</u> variation (geology) may be more important than natural <u>temporal</u> variation (dilution events)

### Why is Electrical Conductivity Important to Aquatic Organisms?

- Electrical conductivity is a measure of salt concentration – as ions, cations, anions
- Ambient salt concentration affects water balance & ion inside the aquatic organism
  - Internal ion balance is important to normal physiological processes
- Potential mechanisms for salt toxicity not clear

### Osmoregulation

- maintaining salt and water balance -



Cells placed in a hypertonic environment shrink due to loss of water. In a hypotonic environment, cells swell due to intake of water. The blood maintains an isotonic environment so that cells neither shrink nor swell. (credit: Mariana Ruiz Villareal)

http://organismalbio.biosci.gatech.edu/nutrition-transport-and-homeostasis/animal-ion-and-water-regulation-i/

### Osmoregulation



Migratory fish must use different physiological mechanisms to survive in (a) freshwater or (b) saltwater environments.



http://organismalbio.biosci.gatech.edu/nutrition-transport-and-homeostasis/animal-ion-and-water-regulation-i/

### Osmoregulation



Most freshwater organisms do not have the ability to move from fresh water to salt water

They can tolerate some increase in salinity, but have limited ability to adapt to great increases in salinity



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Why does salt physiology matter?

### Streams are becoming saltier



Fig. 3. Long-term trends in annual mean water quality measures in the Schuylkill River. Samples were collected from 1973 through 1996 by the USGS and from 1990 through 1999 by PWD. The location for USGS acquired samples is directly adjacent to the Belmont intake, depicted in Fig. 1. All linear trends depicted were statistically significant. Linear slopes and associated statistics are included in the text.

Interlandi & Crockett 2003

### Increased salinization of fresh water in the northeastern United States

Sujay S. Kaushal<sup>\*†‡</sup>, Peter M. Groffman<sup>\*</sup>, Gene E. Likens<sup>\*‡</sup>, Kenneth T. Belt<sup>§</sup>, William P. Stack<sup>¶</sup>, Victoria R. Kelly<sup>\*</sup>, Lawrence E. Band<sup>∥</sup>, and Gary T. Fisher<sup>\*\*</sup>

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Fig. 1. Examples of significant, long-term increases in baseline concentration of chloride for streams and rivers of the northeastern United States. The  $R^2$  values are given for linear regressions. All streams and rivers are located in rural areas but contain roads within their watersheds. (A) LMR0015 (Little Morgan Run), MDE0026 (Middle Run), and BEA (Beaver Run) are sampling stations for tributaries to Liberty Reservoir, a drinking water supply for Baltimore. (B) Wappinger Creek and the Mohawk River are tributaries to the Hudson River in the Hudson River Valley. (C) The streams in the White Mountains drain into Mirror Lake; one is located near an interstate highway in the Hubbard Brook Valley, and the forested reference stream is watershed 6 of the Hubbard Brook Experimental Forest (10).

### History of Road Salt Use in US



Now



Patty Haag WCU MS Student Chester Co Master Watershed Steward







# Salt concentration always elevated, but highly variable





### Toxicity decreases with increased hardness



### Not all species are sensitive



#### Toxicity increases with exposure duration

#### Lethal Concentration (LC50)



### Toxicity is less in cold water



# Salt concentration always elevated, and temperature matters



### Not all salts are equal Acute toxicity (48 h)

	<i>Ceriodaphnia dubia</i> LC50 (mg Cl/L)	<i>Daphnia magna</i> LC50 (mg Cl/L)	<i>Pimephales promelas</i> LC50 (mg Cl/L)
NaCl	1,960	4,770	6,510
MgCl <sub>2</sub>	880	1,330	2,840
CaCl <sub>2</sub>	1,830	2,770	>6,560

### Not all salts are equal Rainbow trout – growth (25 d)



#### Conductivity Correlates with Degradation

157 sites, primarily in Schuylkill River basin



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# Salt Illustrates Land/People Connection to Water:

<u>Toxins:</u> Salt Oils Metals Sealants Herbicides Insecticides Soaps

Drugs Personal Care Fertilizers



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Storm Drains: Sewer System: "Chemical Cocktail" in storm water Freshwater Salinization Syndrome Kaushal et al. 2018

# There are new pollutants of concern today







Winter deicing salts (2003) PAHs in coal tar seal coats (2003)

6PPD-quinone from tires (2020)

### Salt Levels of Concern

LC 50s for a few mayflies moderately hard water at 20°C

	Chloride (mg/L)	Conductivity (uS/cm)
Acute (96 h)	311-2894	1013-8603
Chronic (30 d)	151-898	543-2738

## Points to Remember

- Salt in streams is increasing, especially urban streams
- Elevated salt can cause physiological stress or even be toxic for aquatic organisms. Brief salt spikes in winter are not well understood.
- Impact of elevated salt depends on concentration, stream hardness, exposure duration, type of salt, water temperature, and species



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